



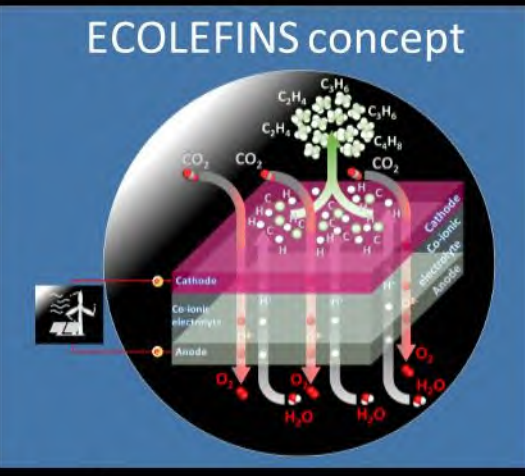
scientific breakthroughs

- H⁺ supply correspond to activities of extremely high pressures
- control H₂ flooding for CH₄ byproduct minimization
- O²⁻ removal to suppress H₂O/CO byproducts and shift conversion
- increase C₂₋₄⁼ selectivity

ECOLEFINS
co-ionics for carbon negative petrochemicals

disconnect light olefins (C₂₋₄⁼) from petroleum

CO₂/H₂O co-electrolysis and CO₂ hydrogenation to C₂₋₄⁼ in **co-ionic Electrochemical Membrane Reactors (ci-EMRs)**



HORIZON-EIC-2022-PATHFINDER-01
101099717 ECOLEFINS
EISMEA funding: 2.5 M€
Duration: 10/23 – 9/26

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CO₂ to C₂₋₄⁼ benchmark (@ 30 bar):
nCO₂ + nH₂ ↔ C_nH_{2n} + 2nH₂O

electrochemically supply protons (H⁺) and remove O²⁻ to/from CO₂

ECOLEFINS' ci-EMR (1 bar):
nCO₂ + 2nH⁺ + 6ne⁻ ↔ C_nH_{2n} + 2nO²⁻



integrate electrolysis endothermicity with CO₂ hydrogenation exothermicity

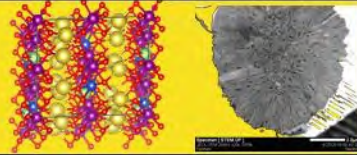
- bypass H₂ compression
- electrolysis at >400 °C
- increase C₂₋₄⁼ yield
- reduce energy demand

reverse 5 – 11 tn CO₂/tn C₂₋₄⁼ emissions to -3 tns CO₂/tn C₂₋₄⁼ capture

innovations

integrated approach

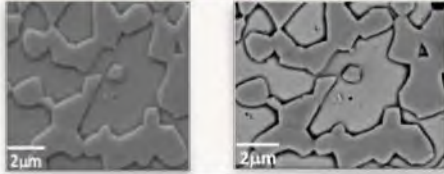
DFT designed
co-ionic membranes and
electrified interfaces



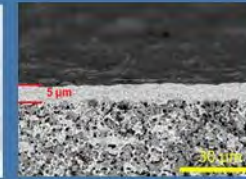
Co-ionic Electrochemical
Membrane Reactors (ci-EMRs)
up-scaling

ci-EMR
tubular
modules

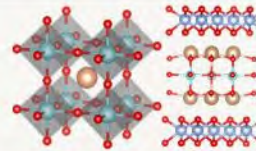
dual phase (cer-cer)
membranes of
co-ionic conductivity



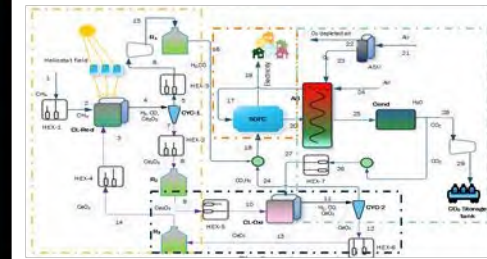
ci-EMR
planar
stacks



electrodes of triple
phase (H^+ , O^{2-} , e^-)
conductivity



planar stacks and
tubular modules testing

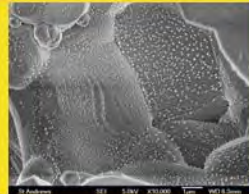
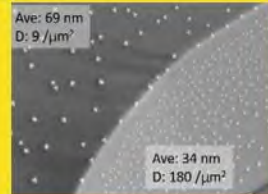


Integrated ECOLEFINS
process modelling

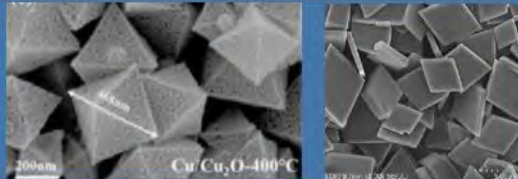
CFD
modelling



exsolutions, for confided and
dispersed
catalytic nanoparticles



MOF-derived
(bi)metallic,
catalytic nanoparticles

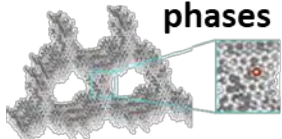


economic feasibility and
sustainability assessment

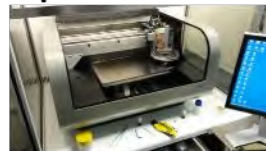


LCSA and
social-LCA

COF-derived
conductive graphitic
phases



inkjet printing for
planar stacks



public awareness
and social acceptance

market opportunities
of ECOLEFINS processes,
ceramics and
electrochemical
components